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Isolation, Characterization and Quantity Determination of Aristolochic Acids, Toxic Compounds in *Aristolochia bracteolata* L.

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Abstract:

Background: Aristolochic Acids (AAs) are major components of plants in *Aristolochia* and have been found to be nephrotoxic, carcinogenic and mutagenic. Herein reported are the isolation, identification and quantity determination methods of Aristolochic Acid-I (AA-I) and Aristolochic Acid-II (AA-II) toxic compounds of *Aristolochia bracteolata* indigenous to Central Sudan and medicinally used in diverse biological functions including analgesic and diuretic effects, treatment of tumors, malaria and/or fevers.

Methods and results: AAs mixture was extracted with methanol from the defatted material of *Aristolochia bracteolata* whole plant at room temperature and was isolated from the aqueous methanol extract by chloroform. Moreover, Silica-gel column chromatography and Preparative Thin Layer Chromatography (PTLC) using chloroform/methanol gradient mixtures were used to isolate AAs mixtures as a yellow crystalline solid. A preliminary detection of AAs was made by Thin Layer Chromatography (silica-gel, chloroform: methanol (6:1)). The R_f value of the acids mixture was 0.43–0.46. The presence of AAs in plant sample was confirmed by High Performance Liquid Chromatography/Ultraviolet (HPLC/UV) analysis using 1% acetic acid and methanol (40:60) as mobile phase and maximum absorption wave length of 250 nm. Quantitative determination of AA-II (49.03 g/kg) and AA-I (12.98 g/kg) was also achieved by HPLC/UV.

Recommendation: It is recommended that the use of *Aristolochia bracteolata* as a medicinal plant should be extremely limited or strictly prohibited. The chromatograms obtained in this study can serve as fingerprints to identify AAs in plant samples.

Keywords: Aristolochic acid-I, Aristolochic acid-II, *Aristolochia*, *Aristolochia bracteolata*

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Introduction

The genus *Aristolochia* grows over a wide area of different climatic zones and consists of about 500 species.¹ Various *Aristolochia* species have been used in herbal medicines since antiquity in obstetrics and in treatment of snakebite,² festering wounds, and tumors, and they remain in use particularly in Chinese herbal medicine.^{3,4} *Aristolochia bracteolata* is known as “worm killer” due to supposed antihelminthic activity and trypanocidal effect.⁵ *Aristolochia bracteolata* also possess a potent antiallergic activity⁶ and has pronounced antibacterial and antifungal activities.⁷

In Sudan, *Aristolochia* is traditionally used as an analgesic, antiscorpion, and antisnake. It is also used in the treatment of tumors, malaria and for fevers.⁸

Aristolochic acids (Fig. 1) are compounds of nitrophenanthrene carboxylic acids that occur naturally in plants of the family Aristolochiaceae, primarily in the genera *Aristolochia* and *Asarum*^{9,10} as well as in butterflies that feed on such plants.¹¹ Botanical products from plants containing aristolochic acids are used in traditional folk medicines, particularly in Chinese herbal medicine, especially those used as a part of weight-loss regimens.⁴

Aristolochic acids are known to be toxic and a rodent carcinogen, in addition to their carcinogenicity, aristolochic acids are also highly nephrotoxic agents.^{1,10–12} In the present work *Aristolochia bracteolata* grown in Sudan was investigated for the presence of Aristolochic acids.

Materials and Methods

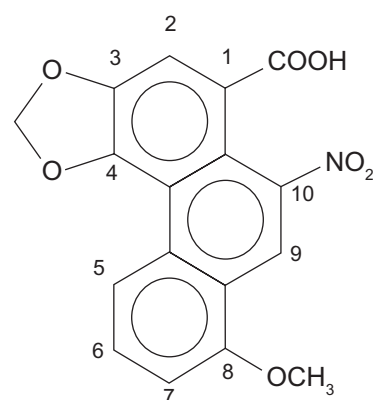
Plant material

The plant *Aristolochia bracteolata* was collected during December 2009 from the Southern Gezira scheme, Sudan. The plant was taxonomically identified in the Department of Chemistry and Pharmacognosy, University of Gezira, and a voucher specimen was deposited.

Methods

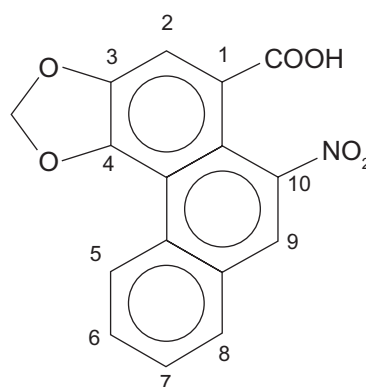
Extraction of plant material

The shade dried whole plant sample (200 g) of *Aristolochia bracteolata* were milled in the form of a coarse powder. The powdered sample was defatted by maceration at room temperature using petroleum ether (40–60 °C) for 6 hours to yield dark green oily


 $C_{17}H_{11}NO_7$

Rel. mol. mass: 341.27

Aristolochic acid I (8-methoxy-6-nitrophenanthro [3, 4-*d*][1, 3] dioxole-5-carboxylic acid).


 $C_{16}H_9NO_6$

Rel. mol. mass: 311.25

Aristolochic acid II (6-nitrophenanthro [3, 4-*d*]-1, 3-dioxole-5-carboxylic acid).

Figure 1. Chemical structure of aristolochic acid I and aristolochic acid II.

residue after evaporation of the solvent. As described in previous studies,^{13,14} and with some modifications (defatting before methanol extraction), the dried defatted residue was macerated in methanol at room temperature for 72 hours and filtered. Solvent was evaporated to dryness under reduced pressure using a rotary evaporator at 60 °C to give a dark green methanol extract.

Liquid-liquid extraction

Twenty grams of *Aristolochia bracteolata* methanol extract was partitioned between water (450 ml) and chloroform (800 ml). Chloroform layer was separated and dried by rotary evaporator at 40 °C to give chloroform extract.

Column chromatography

Chloroform extract of *Aristolochia bracteolata* (4.0 g) was chromatographed on a silica-gel column. For column fractionation the chloroform extract was dissolved in 20 ml chloroform and applied to the top of a column (8.0 cm ID) backed with 160 g silica-gel of a mesh size of 60–120.

Elution was commenced with gradient chloroform/methanol mixture (8:1). Polarity of the solvent was increased in a gradient of 8:1, 4:1, 1:1, 0:1 and finally the column was washed with 100 ml of water. 200 ml of eluents were used and fractions of 50 ml were collected and thin layer chromatography (TLC) monitored. Similar fractions were combined to give three fractions 1–7 (a), 8–14 (b) and 15–17 (c). Fraction (b) was further subjected to column chromatography (3.5 cm ID) backed with 40 g silica-gel of mesh size 60–120, and eluted with a chloroform/methanol mixture of increasing polarity as: 8:1, 4:1, 1:1, 0:1. Fractions of 25 ml were collected and TLC monitored using chloroform/methanol mixture (6:1) as mobile phase (Fig. 2).

Thin layer chromatography (TLC)

TLC glass backed plates were prepared using silica-gel G type 60 (s d Fine-Chem Limited, Mumbai, India). A slurry of silica-gel in water was applied by means of a spreader to glass plates, previously

cleaned with methanol. A layer 0.25 mm thick was used routinely. Plates were activated by heating at 100–110 °C for an hour. TLC aluminum sheets 20 × 20 cm, 1 mm thick, silica-gel 60 F₂₅₄ (Merk, Germany) and Cellulose plastic papers 20 × 20 cm, 1 mm thick (Macherey-Nagel, Germany) were also used. Different mobile phases chloroform/methanol (6:1);¹⁵ chloroform/methanol/acetic acid (65:20:5);¹⁶ benzene/acetone (14:1),¹⁷ chloroform/methanol/water (60:40:10)¹⁸ were checked for the separation of aristolochic acids.

Chromatographic responses (Table 1) were obtained and compared to those of Aristolochic acids reference sample Aristolochic acid mixture of AA-I and AA-II (Aldrich Chemical Company, USA). Day light, ultraviolet light (UV), 0.5% Diphenylamine in 60% H₂SO₄ (DPA/H₂SO₄) and Dragendurff's reagents were used as locating agents to disclose the aristolochic acids on TLC and preparative TLC chromatograms (Fig. 2).

Preparative thin layer chromatography (PTLC)

The dried material obtained from the combined column chromatography fractions which are rich in aristolochic acids (Fraction b) was dissolved in 10 ml chloroform/methanol mixture (6:1) and subjected to preparative thin layer chromatography (PTLC) using 20 × 20 cm cellulose papers for TLC with chloroform/methanol/water (6:4:1) as mobile phase and TLC aluminum sheets: (20 × 20 cm), silica-gel 60 F₂₅₄ with chloroform/methanol (6:1) as mobile phase. The disclosed bands of aristolochic acids were then scraped off, dissolved in chloroform/methanol mixture (6:1) and filtered. The solvent was removed by evaporation and the obtained isolate containing aristolochic acids was subjected to further PTLC for purification using silica-gel fluorescent TLC aluminum sheets as described above (Figs. 3 and 4). Scraped Aristolochic acids were dissolved and filtered. Solvent was removed to yield a yellow crystalline aristolochic acids mixture.

C18-High-performance liquid chromatography (HPLC)

The separation and quantification of aristolochic acids was achieved on HPLC chromatograph, (Waters® 2996, USA); C18 250 × 4.6 mm column with UV detector. A mixture of 1% acetic acid and methanol

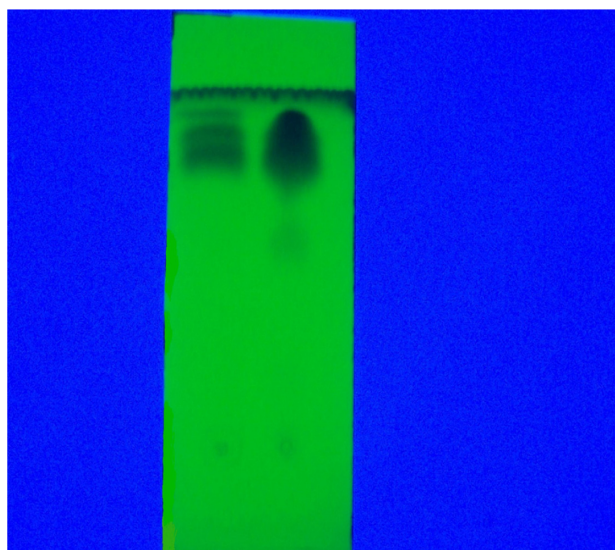


Figure 2. Isolated Aristolochic acids (left), silica-gel fluorescence TLC aluminum sheet, chloroform/methanol (6:1) mobile phase (under UV at 365 nm), compare to standard Aristolochic acids (right).

Table 1. Observed chromatographic responses of Aristolochic acids on TLC chromatograms.

Type of adsorbent	Solvent system	Rf value	Appearance of spots/bands		
			Day light	UV at 365 nm	0.5% DPA/60% H ₂ SO ₄
Silica-gel 60 F ₂₅₄ , precoated aluminum sheets, 20 × 20 cm (Merk, Germany)	CHCl ₃ /MeOH (6:1)	0.43	Yellow fluorescence	Black spot	Pale grey spot
Silica-gel G glass plates (s d Fine-Chem., Limited, Mumbai, India)	CHCl ₃ /MeOH (6:1)	0.46	Yellow fluorescence	Faint brown spot	Pale grey spot
Cellulose, precoated plastic sheets, 20 × 20 cm (Macherey-Nagel, Germany)	CHCl ₃ /MeOH/Water (6:4:1)	0.83	Yellow fluorescence	Faint brown spot	Pale grey spot

(40:60) utilized as mobile phase, flow rate 1.0 ml/min, pressure 3015 psr, temperature 20–25 °C, UV detection 250 nm, Helium degassing, and injection volume 20 µL.

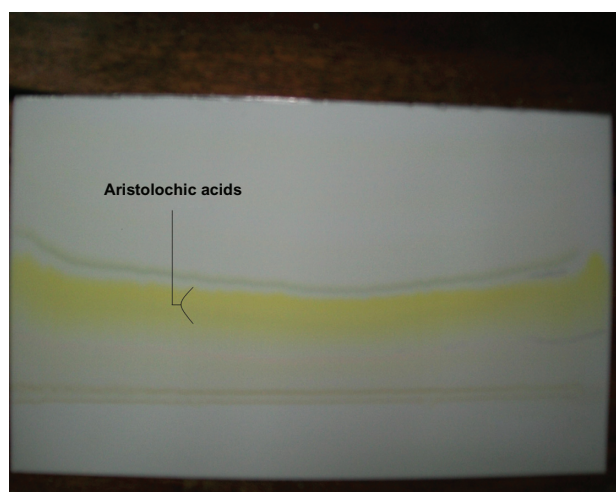
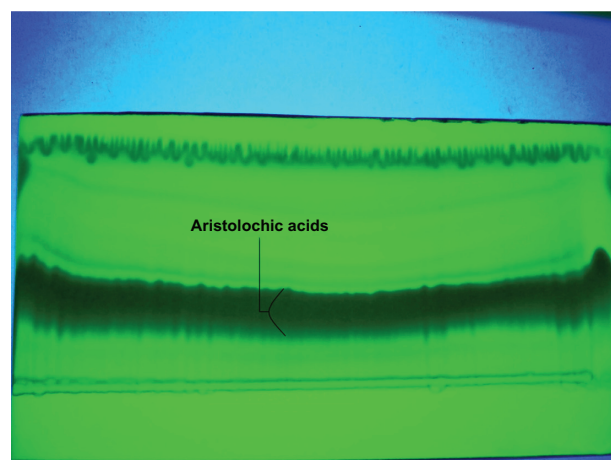
Preparation of standard Aristolochic acids, isolated Aristolochic acids and methanol extract of *Aristolochia bracteolata*

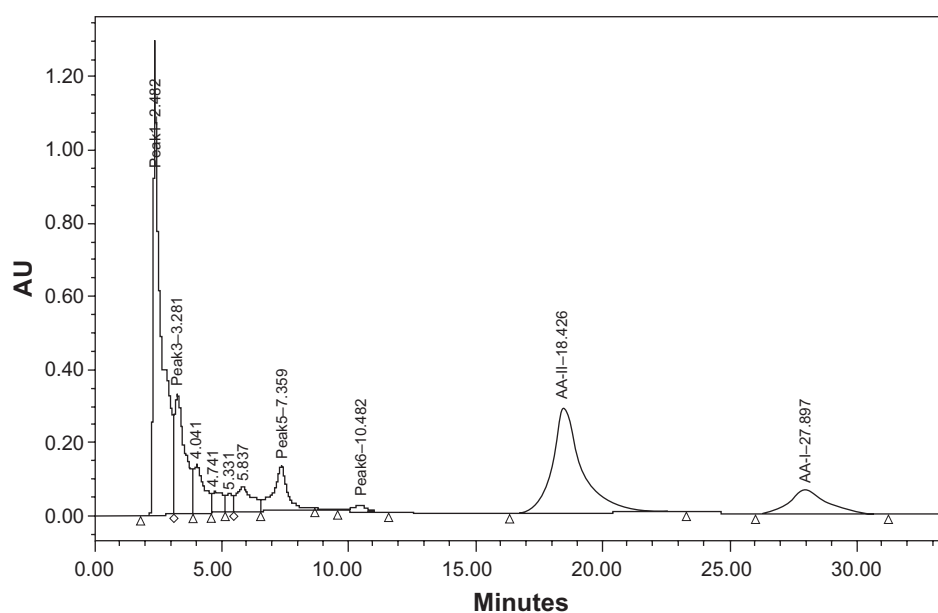
Fifteen milligrams of Aristolochic acids mixture of AA-I and AA-II (Aldrich Chemical Company, USA), fifteen mg of isolated aristolochic acids of *Aristolochia bracteolata* and one g of methanol extract of *Aristolochia bracteolata* were weighed separately using digital balance, each was placed in a small beaker, transferred into volumetric flask (50 ml), washed by methanol-HPLC grade and adjusted to 50 ml, filtered and sonicated for 10 minutes. 20 µL were separately

injected in the HPLC system. Identification was made by comparing the samples retention times with those of the standard reference (Figs. 5–7).

Results and Discussion

To isolate and identify Aristolochic acids (AAs) in *Aristolochia bracteolata* L., the whole plant (200 g) was defatted to yield dark green oily residue (5.35%). As described in previous studies^{13,14} and with some modifications (defatting before methanol extraction, further n-butanol was not used in successive partitioning), the defatted plant material produced on drying 16.5% crude dark green extract on subsequent methanol maceration. On column chromatography of chloroform fraction obtained from the aqueous methanol extract, fraction (b) eluted by 4:1 and 1:1 solvent mixture was found to be rich in Aristolochic acids as

**Figure 3.** Aristolochic acids PTLC chromatogram, silica-gel fluorescence TLC aluminum sheet, chloroform/methanol (6:1) mobile phase, (day light).**Figure 4.** Aristolochic acids PTLC chromatogram, silica-gel fluorescence TLC aluminum sheet, chloroform/methanol (6:1) mobile phase, (under UV at 365 nm).



	Peak Name	RT	Area	% Area	Height
1	Peak1	2.402	26796089	32.59	1292511
2	Peak2	3.029			
3	Peak3	3.281	9595634	11.67	330078
4		4.041	3932822	4.78	134660
5		4.741	1810100	2.20	56422
6		5.331	963237	1.17	49228
7		5.837	3009459	3.66	69800
8	Peak4	6.933			
9	Peak5	7.359	4581045	5.57	119377
10	Peak6	10.482	632544	0.77	17826
11	AA-II	18.426	24433508	29.72	289625
12	AA-I	27.897	6469776	7.87	63686

Figure 5. HPLC chromatogram of methanol extract sample of *Aristolochia bracteolata*.

Condition: column, C18 (Varian® microsb-MV 300-8 C18 250 × 4.6 mm), mobile phase, 1% acetic acid/methanol (40:60), temperature 20–25 °C, flow rate 1.0 ml/min, UV detection 250 nm.

revealed by TLC monitoring. Column chromatography fractionation of fraction (b) with the same gradient mobile phase and TLC monitoring showed that Aristolochic acids were obtained as major spots in the first ten fractions. Sixteen milligrams of a yellow powder of Aristolochic acids mixture (Fig. 2) was produced thereafter by preparative thin layer chromatography.

It was noted in previous studies that many attempts have been made to assay active components in Aristolochiaceae by various methods including Thin Layer Chromatography, High Performance Liquid Chromatography-Ultraviolet. However, most of these

methods focus on analysis of AA-I and AA-II in samples.^{11,14,15,19,20}

In the present study TLC and PTLC chromatograms clearly demonstrated the presence of Aristolochic acids. Results also indicated that the obtained AAs produced varying chromatographic responses with different chromatographic solvent systems and/or locating agents (Table 1). Aristolochic acids generally acquired black color on fluorescent silica-gel in UV-light at 365 nm while they fluoresce yellow in day light. Rf values ranged 0.43–0.46 (chloroform/methanol (6:1); silica-gel) and 0.83 (chloroform/methanol/water (6:4:1); cellulose).

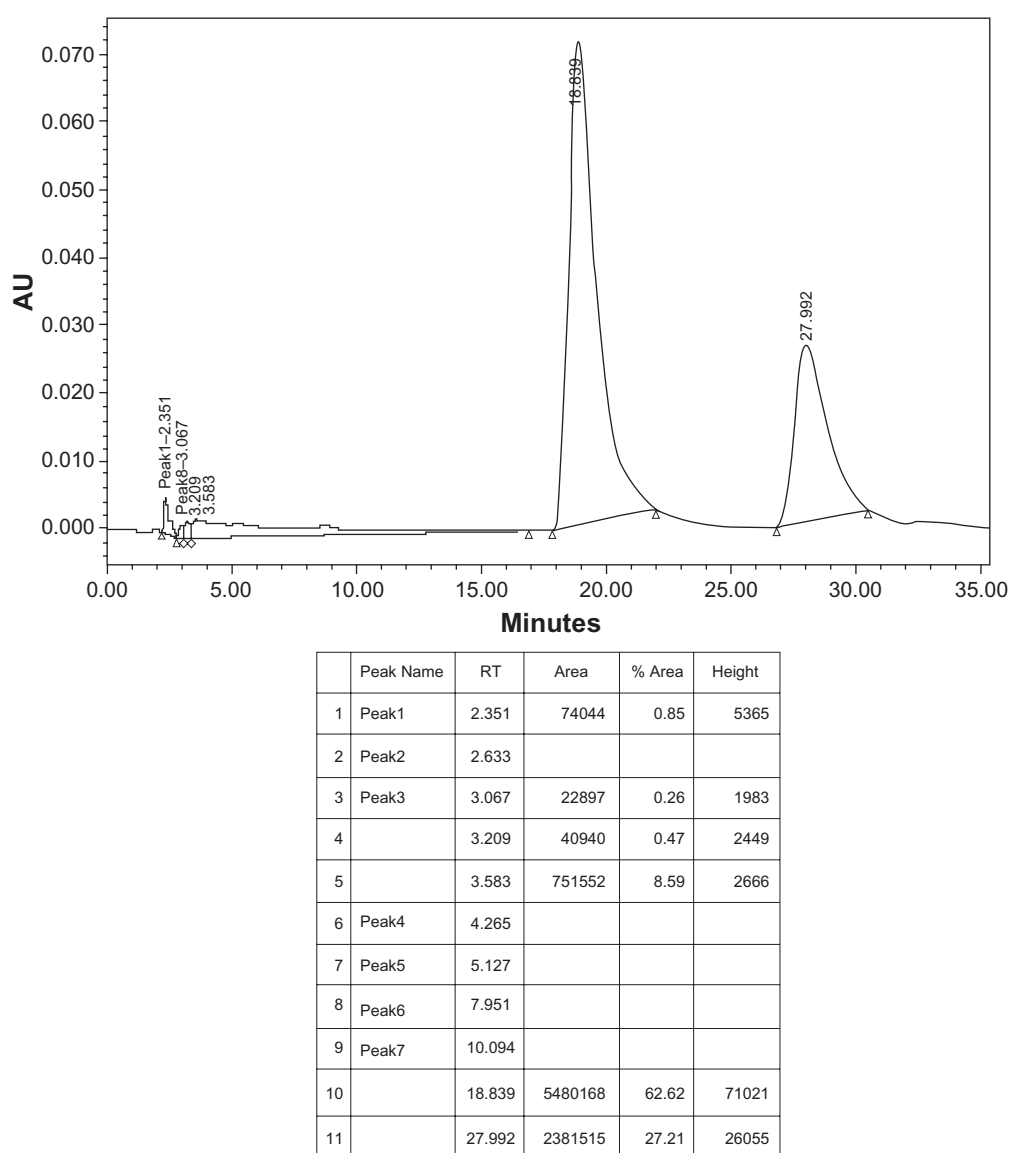


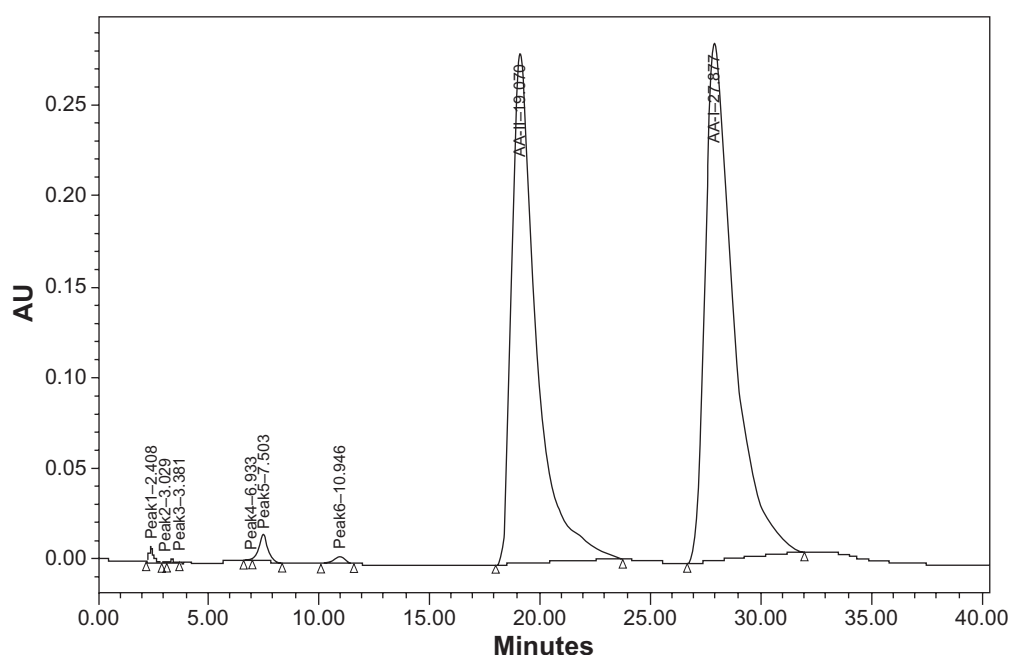
Figure 6. HPLC chromatogram of isolated AAs sample of *Aristolochia bracteolata*.

Condition: column, C18 (Varian® microsb-MV 300-8 C18 250 × 4.6 mm), mobile phase, 1% acetic acid/methanol (40:60), temperature 20–25 °C, flow rate 1.0 ml/min, UV detection 250.

For separation and quantification of Aristolochic acids by HPLC earlier reports mainly used a mobile phase of methanol and water^{21–24} or acetonitrile and water^{21,25,26} and the optimum chromatographic conditions were investigated by varying the content of methanol and water.

In this study, and for the first time in Sudan, the presence of Aristolochic acids in *Aristolochia bracteolata* methanol extract was confirmed and quantified by High Performance Liquid Chromatography as shown in Figures 5–7 using a mixture of 1% acetic acid and methanol (40:60) as mobile phase. Aristolochic acid II (AA-II) and Aristolochic acid I (AA-I) were detected as

major peaks of components at a retention time of about 18.4 min and 27.9 min respectively and exactly corresponding to data profile exhibited by the Aristolochic acid reference sample. HPLC data also showed that AA-II was represented in a higher calculated amount of 49.03 g/kg compared to AA-I (12.98 g/kg) in *Aristolochia bracteolata* L. whole plant while in literature, other species of *Aristolochia* had recorded far more lower amounts of AAs such as *Aristolochia debilis* that contained 0.18 g/kg of AA-II and 1.01 g/kg of AA-I; *Aristolochia manshuriensi* contained 1.0 g/kg of AA-II and 8.82 g/kg of AA-I; *Aristolochia fangchi* also contained 0.22 g/kg of AA-II and 2.22 g/kg of AA-I.^{1,11,27–32}



	Peak Name	RT	Area	% Area	Height
1	Peak1	2.408	133058	0.28	8675
2	Peak2	3.029	15311	0.03	1354
3	Peak3	3.361	38589	0.08	2092
4	Peak4	6.933	15507	0.03	1391
5	Peak5	7.503	407054	0.85	14153
6	Peak6	10.946	156466	0.33	4078
7	AA-II	19.078	21782025	45.33	281529
8	AA-I	27.877	25503012	53.07	286251

Figure 7. HPLC chromatogram of AAs standard sample.

Condition: column, C18 (Varian® microsb-MV 300-8 C18 250 × 4.6 mm), mobile phase, 1% acetic acid/methanol (40:60), temperature 20–25 °C, flow rate 1.0 ml/min, UV detection 250 nm.

It is worth noting that, in the reported data AA-I obtained from different *Aristolochia* species was found to be in a higher quantity than AA-II, contrary to results obtained in this study. This may be attributed to climatic or biological variations which could affect metabolic pathways of interconversion of Aristolochic acids and analogues.

In conclusion, the presence of Aristolochic acids toxic compounds in *Aristolochia bracteolata* L. methanol extract was evident through the multi-component method including Column Chromatography (CC), Thin Layer Chromatography (TLC), Preparative Thin Layer Chromatography (PTLC) and had been confirmed and quantified by High Performance Liquid Chromatography (HPLC).

Thus it is recommended that the use of *Aristolochia bracteolata* as a medicinal plant should be extremely limited or strictly prohibited. Moreover, the chromatograms obtained in this study can serve as fingerprints to identify plants suspected to contain or adulterated with Aristolochic acids.

Disclosure

This manuscript has been read and approved by all authors. This paper is unique and is not under consideration by any other publication and has not been published elsewhere. The authors and peer reviewers of this paper report no conflicts of interest. The authors confirm that they have permission to reproduce any copyrighted material.



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