

Effects of Magnesium and Calcium on the Oxygenation Reaction of Erythrocruorin from the Marine Polychaete Arenicola marina and the Terrestrial Oligochaete Lumbricus terrestris

Authors: Ochiai, Takehiko, and Weber, Roy E.

Source: Zoological Science, 19(9): 995-1000

Published By: Zoological Society of Japan

URL: https://doi.org/10.2108/zsj.19.995

The BioOne Digital Library (<u>https://bioone.org/</u>) provides worldwide distribution for more than 580 journals and eBooks from BioOne's community of over 150 nonprofit societies, research institutions, and university presses in the biological, ecological, and environmental sciences. The BioOne Digital Library encompasses the flagship aggregation BioOne Complete (<u>https://bioone.org/subscribe</u>), the BioOne Complete Archive (<u>https://bioone.org/archive</u>), and the BioOne eBooks program offerings ESA eBook Collection (<u>https://bioone.org/esa-ebooks</u>) and CSIRO Publishing BioSelect Collection (<u>https://bioone.org/csiro-ebooks</u>).

Your use of this PDF, the BioOne Digital Library, and all posted and associated content indicates your acceptance of BioOne's Terms of Use, available at www.bioone.org/terms-of-use.

Usage of BioOne Digital Library content is strictly limited to personal, educational, and non-commercial use. Commercial inquiries or rights and permissions requests should be directed to the individual publisher as copyright holder.

BioOne is an innovative nonprofit that sees sustainable scholarly publishing as an inherently collaborative enterprise connecting authors, nonprofit publishers, academic institutions, research libraries, and research funders in the common goal of maximizing access to critical research.

Effects of Magnesium and Calcium on the Oxygenation Reaction of Erythrocruorin from the Marine Polychaete Arenicola marina and the Terrestrial Oligochaete Lumbricus terrestris

Takehiko Ochiai^{1*} and Roy E. Weber²

¹Department of Biology, Fukushima Medical University School of Medicine, Fukushima 960-1295, Japan ²Department of Zoophysiology, University of Aarhus, DK-8000 Aarhus C, Denmark

ABSTRACT—Oxygenation function of annelid erythrocruorin (Er) is affected by Mg and Ca concentration in the blood. Four classes of responces may be encountered in different species: 1) Mg=Ca (equal effects), 2) Mg>Ca, 3) Mg<Ca and 4) no effect. In the marine polychaete *Arenicola marina*, at physiological pH and 20°C, Mg and Ca exerted almost equivalent effects in increasing oxygen affinity in the range of 1–200 mM. As measured from the slope of $\Delta \log P_{1/2} / \Delta \log$ [Cation] the effect of Mg was larger than that of Ca at the physiological concentration of respective ion (55 mM Mg; 10 mM Ca). The n_{1/2} value was similar in the presence of both cations (pH 7.0) or higher for Mg (pH 7.6). In the terrestrial oligochaete *Lumbricus terrestris*, at the same condition, Ca was more effective than Mg, in raising oxygen affinity at both pHs, also at the physiological concentration (2–4 mM Mg; 8 mM Ca), and n_{1/2} was similar for Mg and Ca (pH 7.0) or higher for Ca (pH 7.6). The Bohr factor, -[$\Delta \log P_{1/2} / \Delta pH$], is maintained its maximum value within the span of the physiological concentration of Mg in *Arenicola*. In *Lumbricus*, Ca can contribute to the increase of the Bohr factor at the physiological concentration, but Mg cannot contribute to it. These results reveal that *Arenicola* and *Lumbricus* belong to classes 1) and 3), respectively, and that the oxygenation function of both Ers may be controlled by effective utilization of the more dominant of the divalent cations Mg and Ca.

Key words: annelids, erythrocruorin, oxygenation, Bohr effect, Mg/Ca effect

INTRODUCTION

In contrast to the vertebrate hemoglobin, where oxygen affinity is decreased by organic phosphates packaged in the red blood cells, the affinity of the high molecular-weight extracellular annelid respiratory proteins, termed either erythrocruorins (Ers) or hemoglobins, is increased by inorganic divalent cations. The effect of cations does not seem simply to be a charge effect since the effect of divalent cations is much greater than that of monovalent cations at double that concentration (Everaarts and Weber, 1974). Recent data suggest marked differences in the sensitivities of Er from different classes of annelids to Mg and Ca (Weber and Vinogradov, 2001). In marine polychaetes, the sensitivity shows much variation. In *Perinereis aibuhitensis*, Mg is more effective than Ca (Tsuneshige *et al.*, 1989). In *Are*-

* Corresponding author: Tel. +81-24-548-2111; FAX. +81-24-548-3836. E-mail: tochiai@fmu.ac.jp

nicola marina, Mg and Ca exerts the same effect (Everaarts and Weber, 1974). In Amphitrite ornata, Ca is rather effective than Mg (Chiancone et al., 1981). In Marphysa sanguinea, Er appears to be insensitive to Mg and Ca (Imai et al., 1990), as is observed with Eurythoe complanata Er and Mg (llan et al., 1990). In oligochaetes, on the other hand, Ca is more effective than Mg in almost all species examined, as Eisenia foetida (Ochiai, 1984; Igarashi et al., 1985; Igarashi et al., 1991), Lumbricus terrestris (Fushitani et al., 1986), and Pheretima hilgendorfi (Ochiai et al., 1993). However, in Glossoscolex paulistus, Mg and Ca exert the same effect on the affinity at the physiological range (Margues and Meirelles, 1995). As an exception, the aquatic species Tubifex hattai shows no response to the divalent cations (Ochiai et al., 1991). In leech, Ilan and Harown (1993a,b) reported greater sensitivity to Ca than Mg in Hirudo medicinalis and Macrobdella decora, however, similar sensitivities to both cations at 50mM were observed in M. decora (Weber et al., 1995). The effect of divalent cations may be physiologically important in osmoconforming euryhaline animals like Are*nicola* (Krogh-Rasmussen and Weber, 1979). Interaction with divalent cations is further of interest since it may change oxygen affinity of Er through alterations in proton concentrations at the molecular surface (Weber and Olsen, 1984).

The main aim of this study is to investigate differences in the sensitivities to Mg and Ca in annelids, and to discuss the relation between the sensitivity difference and ionic composition in the blood.

MATERIALS AND METHODS

Materials and preparations

The polychaete annelid, Arenicola marina, was obtained from Esbjerg on the west coast of Jutland, Denmark. Blood was collected as previously described (Krogh-Rasmussen and Weber, 1979). Blood was centrifuged at 14.000 rpm for 30 min to remove particulate matter, dialysed overnight against 100 fold excess of 0.01 M Tris-HCl buffer, pH 7.6 that was exchanged 4 times, and again centrifuged at 14,000 rpm for 30 min. The oligochaete annelid, Lumbricus terrestris, was collected in Aarhus and Frederikshavn, Denmark, and was identified according to the description of Sims and Gerard (1985). Blood was collected in glass hematocrit tubes as described for Pheretima hilgendorfi (Ochiai, 1983), centrifuged at 12,000 rpm for 5 min, pooled and diluted approximately 25 fold with 0.01 M Tris-HCl buffer, pH7.6, containing 0.5 mM PMSF as a proteolysis inhibitor (James, 1978). The Er was then precipitated by ultracentrifugation at 150,000 g for 1 hr (Beckman L8-70M, type 65 rotor) and gently resuspended in the same buffer without PMSF. After an additional centrifugation the pellet was suspended in a small amount of the buffer. Particulate matter was then removed by spinning at 21,800 g for 20 min. The Er samples were divided in 0.1 ml aliquots and were frozen at -80°C.

Oxygen equilibrium experiment

The aliquots were thawed individually on the same day as the oxygen equilibrium determinations were carried out. Before measurements, accurate volumes of standard cation solutions and 1 M BisTris propane or Tris buffers were added in the order and by the addition of distilled water to give differing cation levels, a final buffer concentration of 0.1 M and the same Er concentration in the individual tubes. All cations were added as chloride salts. Cation levels were assayed by measuring chloride concentrations using a Radiometer CMT 10 chloride titrator, and by atomic absorption spectroscopy. pH values were measured using a microelectrode set-up (Radiometer BMS 2 Mk 2 coupled to PHM64). Er concentrations were estimated from absorbances at 540 nm, using a millimolar absorption coefficient of 14.37 for human oxyhemoglobin (Assendelft, 1970), and methemoglobin contents were estimated by the spectral change of oxyhemoglobin at 574 nm. Oxygen equilibria were recorded with diffusion chamber modified after Sick and Gersonde (1969). Final heme concentrations were 1.4-1.5 mM in Arenicola Er and 0.8-1.1 mM in Lumbricus Er.

RESULTS AND DISCUSSION

Figs. 1 and 2 show the effects of Mg and Ca on the oxygen equilibrium curves of Er from *Arenicola marina* at pH 7.0 and 7.6 and of Er from *Lumbricus terrestris* at pH 7.0 and 7.8 on the same scale. At pH 7.0, Mg and Ca levels of 10 mM exerted no effect on the oxygenation curve of *Arenicola* Er (Fig. 1), whereas 150–170 mM Mg and Ca increased oxygen affinity. At pH 7.6, 9–13 mM Mg and Ca slightly increased oxygen affinity, whereas more marked increases



Fig. 1. Effects of Mg and Ca on the oxygen equilibrium curves of Er from *Arenicola marina*. Buffer: 0.1 M BisTris propane at pH 7.0 (6.98–7.01) and 0.1 M Tris at pH 7.6 (7.58–7.70), 20°C. pH 7.0: — —, no additive (1.88 mM Mg, 1.03 mM Ca); , 8.20 mM Mg; , 11.7 mM Ca; , 149 mM Mg; , 166 mM Ca. pH 7.6: -- 〇 --, no additive (1.60 mM Mg, 1.06 mM Ca); , 9.30 mM Mg; , 12.9 mM Ca; , 157 mM Mg; , 129 mM Ca.



Fig. 2. Effects of Mg and Ca on the oxygen equilibrium curves of Er from *Lumbricus terrestris*. Buffer: 0.1 M BisTris propane at pH 7.0 (6.99–7.04) and 0.1 M Tris at pH 7.8 (7.72–7.80), 20°C. pH 7.0: — —, no additive (0.20 mM Mg, 1.70 mM Ca); , 9.60 mM Mg; , 12.4 mM Ca; , 160 mM Mg; , 170 mM Ca. pH 7.8: -- \bigcirc --, no additive (0.20 mM Mg, 1.55 mM Ca); , 9.50 mM Mg; , 11.8 mM Ca; , 170 mM Mg; , 162 mM Ca.

were induced by 130–160 mM Mg and Ca. At both pH values, no significant difference was found between the effects of these two cations. In *Lumbricus* (Fig. 2), 10–12 mM Mg and Ca did not give a remarkable effect at pH 7.0, but def-

inite increase of oxygen affinity was observed at pH 7.8. The effect of Ca was obviously larger than that of Mg especially at pH 7.8. At both pHs, 160-170 mM Mg and Ca induced marked increase in oxygen affinity, the effect of Ca was somewhat larger than that of Mg. When compared with the oxygenation curve in the absence of the cations and that in the presence of the cations, both curves overlapped with each other at low oxygen saturation, but did not overlap at high saturation, and this phenomenon was more evident in Lumbricus than in Arenicola. This result implies preferential binding of both cations to oxygenated high affinity R (relaxed) state form of Arenicola and Lumbricus Ers, and the existence of at least two high affinity R states (Weber, 1981; Weber and Vinogradov, 2001). However, a resonance Raman spectroscopic study of Lumbricus deoxygenated low affinity T (tense) state and oxygenated R state showed no spectral changes involving the iron-proximal histidine stretching mode (Vidugiris et al., 1993). Moreover, small angle X-ray scattering experiments of Lumbricus Er revealed that the overall volume almost unaltered upon deoxygenation (Krebs et al., 1996).

Figs. 3 and 4 illustrate the concentration dependence of the effects of the cations on $P_{1/2}$ (oxygen tention at half-saturation) and $n_{1/2}$ (cooperativity coefficient at half-saturation) at pH 7.0 and 7.6/7.8 in Arenicola and Lumbricus, respectively. As evident from Fig. 3a, Mg and Ca exerted identical effects in increasing oxygen affinity of Arenicola Er at both pH values. At pH 7.0, increasing concentrations of both Mg and Ca raised n_{1/2} values, from about 3.5 to 4.5. At pH 7.6, n_{1/2} was almost unaffected at about 3.5 by Mg at all concentrations tested, and by Ca at concentrations below approximately 30 mM. At higher Ca levels n_{1/2} fell to 2.5-3.0 (Fig. 3b). In Lumbricus, Ca was more effective than Mg in raising oxygen affinity at both pH 7.0 and 7.6, and the difference between the Mg and Ca effects were somewhat larger at pH 7.6 than pH 7.0 (Fig. 4a). As regards $n_{1/2}$, at pH 7.8 the value was almost constant at about 6-7 at Mg and Ca concentrations up to 50 mM, then gradually decreased to reach 3-4 at concentrations near 500 mM. Near 10 mM, however, Ca gave somewhat higher $n_{1/2}$ value than Mg. At pH 7.0, the value was 2.8 up to 10 mM Ca and Mg and then increased to 3.2-4.1 at higher cation levels (Fig. 4b). n_{1/2} at pH 7.8 were much higher than those at pH 7.0, especially at low concentration of the cations, which is reverse compared to Arenicola. These results show P1/2 of Arenicola Er has almost the same reactivity to Ca and Mg, whereas Lumbricus Er shows higher reactivity to Ca than to Mg on both pH values tested. Moreover, $n_{1/2}$ of Arenicola Er shows the same or higher reactivity to Mg, whereas Lumbricus Er shows the same or higher reactivity to Ca between the two cations. Another illustration of the cation concentration dependency of $P_{1/2}$ is furnished by a comparison of Δlog P_{1/2} /Δlog [Cation] at 1–200 mM cation level represented by Table 1. It is evident that Lumbricus Er is 1.2 and 1.6 fold sensitive than Arenicola Er for Mg and Ca at pH 7.0, respectively, and 1.7 and 2.0 fold sensitive for each cation at pH



Fig. 3. Mg and Ca concentration dependency of log $P_{1/2}$ (a) and $n_{1/2}$ (b) of Er from *Arenicola marina*. 0.1 M BisTris propane buffer at pH 7.0 (open symbols) and 0.1 M Tris buffer at pH 7.6 (closed symbols), 20°C. (--, --) Mg and (---, --) Ca.



Fig. 4. Mg and Ca concentration dependency of log $P_{1/2}$ (a) and $n_{1/2}$ (b) of Er from *Lumbricus terrestris*. 0.1 M BisTris propane buffer at pH 7.0 (open symbols) and 0.1 M Tris buffer at pH 7.6/7.8 (closed symbols), 20°C. The values of $P_{1/2}$ at pH 7.8 were corrected to those at pH 7.6. $n_{1/2}$ values are those at pH 7.8. (- -, -) Mg and (- - -, - -) Ca.

7.6, respectively.

According to a bracelet model for *Lumbricus* Er proposed by Vinogradov and his collaborators, the hexagonal

Table 1. Cation concentration dependency of $P_{1/2}$ between 1 and 200 mM cation ($\Delta \log P_{1/2}/\Delta \log$ [Cation]) in the Ers from *Arenicola marina* and *Lumbricus trrestris*

	$\Delta log P_{1/2} / \Delta log [Mg]$		$\Delta log \; P_{1/2} / \Delta log \; [Ca]$	
	pH 7.0	pH 7.6	pH 7.0	pH 7.6
Arenicola marina	0.100	0.117	0.100	0.126
Lumbricus terrestris	0.117	0.204	0.161	0.252

bilayer structure is retained 12 dodecamers, consisting of three copies each of disulfide bonded trimer and monomer chains $(12 \times [a + b + c]_3 [d_3])$, which are decorated with 36– 42 linker chains. Chain a and one of the linkers are glycosylated (Vinogradov et al., 1986; Vinogradov et al., 1991; Martin et al., 1996). Arenicola Er shows indistinguishable whole feature from that of Lumbricus Er, however, its proposed dodecamer composition is one trimer with nine monomer chains, six groups of the dodecamer are assembled and the center of which contains two extra copies of trimers. and its two sets are decorated with 42 linker chains. Arenicola Er does not have glycosylated chains in contrast to Lumbricus Er (Zal et al., 1997). On the basis of a molecular model for Lumbricus Er proposed by Taveau et al. (1999), Kuchumov et al. (2000) classified the Ca2+-binding sites of the Er into three groups by a relative affinity when compared to Ca and EDTA. According to the model, the low-affinity type (i) sites of the three, more than 100 sites in number with affinities lower than EDTA, involve the effects of 1-100 mM Group IIA cations on the Er structure and function. Most of the type (i) sites distribute to one of the linker chains (L3) and partly occur within the dodecamer subassembly. Because of the type (i) sites predominate in the linker chain, it is possible to attribute the difference of the Mg/Ca reactivity between both Ers to that of the linker chain content or the participation of glycosilation. However, the content is very similar in both Ers (Martin et al., 1996; Zal et al., 1997), and the L3 chain of Lumbricus Er is not glycosylated (Martin et al., 1996).

Unfortunately, the available information on the Ca²⁺and/or Mg²⁺-binding site of Arenicola Er is limited. Weber and Olsen (1984) observed that the Mg titration of Arenicola Er induced proton release from its molecular surface. In this experiment, a similar decrease in pH was found when Ca was added to the Er solution as well. These facts are an indirect but a clear evidence indicating the presence of Mg²⁺-binding sites and another Ca²⁺-ones for Arenicola Er. In this Er preparation, Arenicola Er retains approximately 170 Mg and 100 Ca per mole, and Lumbricus Er contains 27 Mg and 220 Ca per mole (see legends of Fig. 1 and Fig. 2). Especially in Lumbricus Er, each value is somewhat higher than that of the Ca reported by Standley et al. (1988) and those of the Mg and Ca for Pheretima hilgendorfi (Ochiai et al., 1993), respectively. This may simply be due to the fact that dialysis step was omitted in this preparation, however, their numerical values may indicate the specific affinity of Ca for the Er. On the other hand, Arenicola Er, which includes dialysis in the process of preparation, still retains comparable amount of Mg in addition to Ca. This fact suggests the presence of at least two kinds of binding sites, one for Mg^{2+} and the other for Ca²⁺, for *Arenicola* Er, and that these sites constitute the type (ii), type (iii), and a part of type (i) sites by analogy with *Lumbricus* Er (Kuchumov *et al.*, 2000). Anyway, classification of the binding sites and their identification for *Arenicola* Er must wait for future research.

Very recently, Jouan *et al.* (2001) classified annelid Ers possessing a hexagonal bilayer structure into two groups by cryoelectron microscopic observation. Type-I appears as the vertices of the upper layer are 16° clockwise rotated with respect to those of the lower layer. Type-II has two eclipsed hexagonal layer structure. Type-I occurs in oligochaete like *Lumbricus*, achaete and vestimentifera. Type-II is found in polychaete, the representative is *Arenicola*. As linker mediated equatorial or axual interactions are believed to contribute the high cooperativity of annelid Ers (Lamy *et al.*, 1996), the possibility cannot be excluded that the presence or absence of the 16° rotation participates in the function of both Ers.

Fig. 5 shows the Bohr factors (-[$\Delta \log P_{1/2} / \Delta pH$]) at various concentrations of Mg and Ca observed in the pH 7.0–7.6 in *Arenicola* and pH 7.0–7.8 in *Lumbricus*. In *Arenicola*, the value is highest (near –0.6) at 30–80 mM Mg and Ca concentrations, but decreases both low and high levels of the cation concentrations. In *Lumbricus*, the value begins to increase from –0.4 at 5 mM Mg and Ca, then reaches the maximum value of –0.75 at 40 mM Mg and Ca, and maintains the same value above 40 mM Mg and Ca. Ca concentration in the blood is about 10 mM in both *Arenicola* and *Lumbricus*, while Mg concentration is far higher in *Arenicola* than in *Lumbricus* (55 mM and 2–4 mM, respectively: Kamemoto *et al.*, 1962; De Jorge *et al.*, 1965; De Jorge and



Fig. 5. Effects of Mg and Ca on the Bohr effects of Ers from *Arenicola marina* and *Lumbricus terrestris*. Bohr factors were calculated from the log P1/2 values at pH 7.0 and 7.6 in *Arenicola* (open symbols, broken line), and at pH 7.0 and 7.8 in *Lumbricus* (closed symbols, solid line). (,) Mg and (,) Ca.

Polychaeta		
Mg Ca	Arenicola marina	Everaarts and Weber (1974);
		Present study
Mg > Ca	Perinereis aibuhitensis	Tsuneshige <i>et al</i> . (1989)
Mg < Ca	Amphitrite ornata	Chiancone et al. (1981)
no effect	Marphysa sanguinea	Imai <i>et al</i> . (1990)
Oligochaeta		
Mg Ca	Glossoscolex paulistus	Marques and Meirelles (1995)
Mg < Ca	Eisenia foetida	Ochiai (1984);
		lgarashi <i>et al</i> . (1985, 1991)
11	Lumbricus terrestris	Fushitani <i>et al</i> . (1986);
		Present study
"	Pheretima hilgendorfi	Ochiai <i>et al</i> . (1993)
no effect	Tubifex hattai	Ochiai <i>et al</i> . (1991)
Hirudenea		
Mg < Ca	Hirudo medicinalis	llan and Haroun (1993a)
"	Macrobdella decora	llan and Haroun (1993b)

Table 2. Effects of Mg and Ca on the oxygen affinity of annelid Er

Sawaya, 1967; Prosser, 1973; Oglesby, 1978; Ochiai et al., 1993). Here again we try to compare the reactivity of both Ers to each cation under their physiological concentrations. In Arenicola, the reactivity as measured from the slope of the tangential line of log $P_{1/2}$ vs log [cation] curve at the physiological concentration of respective ion (55 mM Mg; 10 mM Ca) is larger for Mg than that for Ca, especially at pH 7.0 (Fig. 3a). In Lumbricus, the slope for Ca is larger than that for Mg at pH 7.6 but similar at pH 7.0, at physiological concentrations of respective ion (2-4 mM Mg; 8 mM Ca) (Fig. 4a). These results reveal that the oxygen affinities of both Ers may be controlled by effective utilization of the more dominant of the divalent cations Mg and Ca. The Bohr factor may also be under the control of each cation concentration. In Arenicola, the blood concentrations of Mg and Ca decrease sharply upon transfer from sea water to brackish water from about 60 to 20 mM and 8 to 3 mM, respectively (Krogh-Rasmussen and Weber, 1979). However, as evident from Fig. 5, the Bohr factor is maintained its maximum value within the span of the physiological concentration of Mg, so it is considered Mg is made efficient use through the rise of the Bohr factor for the oxygen delivery between blood and tissues. In Lumbricus, as shown in Fig. 5, the effects of Ca and Mg on the rise in Bohr factor is much larger than those of the case of Arenicola at more than 10 mM. However, at physiological concentration of respective ion, Mg cannot contribute to the increase of the Bohr factor, but Ca can contribute to it. For this terrestrial earthworm, the increase of carbon dioxide concentration may be more important than the change of ionic concentration in the milieu on the Bohr effect (Weber and Baldwin, 1985).

The effect of Mg and Ca on the oxygen affinity of annelid Er is summarized in Table 2. Four cases of responses may be encountered in different species: 1) Mg=Ca (equal effects), 2) Mg>Ca, 3) Mg<Ca and 4) no effect. This report revealed that *Arenicola* and *Lumbricus* belonged to classes 1) and 3), respectively. In contrast to marine polychaetes that are represented in all four classes 1) - 4), terrestrial oligochaetes fall into the classes 1) and 3). This difference may relate to the specific habitats of the two animal groups.

ACKNOWLEDGEMENTS

We thank Dr. Ole Mather-Christensen, Zoology Institute, for help and advice in identification of earthworms and Prof. Torben Clausen, Physiology Institute, Aarhus University, for some atomic absorption measurements. This work was supported by the Natural Science Research Council of Denmark, and by a grant-in-aid for Scientific Research (No. 08640871) from the Ministry of Education, Culture and Science of Japan.

REFERENCES

- Assendelft OW (1970) Spectrophotometry of Haemoglobin Derivatives. Royal VanGorcum, Assen, The Netherlands
- Chiancone E, Ferruzzi G, Bonaventura C, Bonaventura J (1981) Amphitrite ornata erythrocruorin II. Molecular controls of function. Biochim Biophys Acta 670: 84–92
- De Jorge FB, Haeser PE, Ditadi ASF, Petersen JA, Ulhôa Cintra AB, Sawaya P (1965) Biochemical studies on the giant earthworm *Glossoscolex giganteus* (Leuckart). Comp Biochem Physiol 16: 491–496
- De Jorge FB, Sawaya MC (1967) Comparative biochemical studies on the oligochaetes *Pheretima hawayana, Glossoscolex grandis*, and *Rhinodrilus* sp. Comp Biochem Physiol 22: 359–369
- Everaarts JM, Weber RE (1974) Effects of inorganic anions and cations on oxygen binding of haemoglobin from *Arenicola marina* (polychaeta). Comp Biochem Physiol 48A: 507–520
- Fushitani K, Imai K, Riggs AF (1986) Oxygenation properties of hemoglobin from the earthworm, *Lumbricus terrestris*. Effects of pH, salts, and temperature. J Biol Chem 261: 8414–8423
- Igarashi Y, Kimura K, Kajita A (1985) Calcium-dependent allosteric

modulation of the giant hemoglobin from the terrestrial oligochaete, *Eisenia foetida*. Biochem Int 10: 611–618

- Igarashi Y, Kimura K, Kajita A (1991) Analysis of oxygen equilibria of the giant hemoglobin from the earthworm *Eisenia foetida* using the Adair model. J Biochem 109: 256–261
- Ilan E, Azem A, Daniel E (1990) Structural characterization and oxygen binding properties of extracellular hemoglobin from the marine polychaete *Eurythoe complanata*. Comp Biochem Physiol 96B: 783–786
- Ilan E, Haroun J (1993a) Oxygen-binding properties of extracellular hemoglobin from the leech, *Hirudo medicinalis*. Effects of pH, cations and temperature. Biochim Biophys Acta 1162: 77–83
- Ilan E, Haroun J (1993b) Physicochemical features and oxygenation properties of extracellular hemoglobin from the leech, *Macrobdella decora*. Comp Biochem Physiol 104B: 833–839
- Imai K, Hori H, Gotoh T (1990) Physiological and biochemical properties of the hemoglobin from *Marphysa sanguinea*. Seikagaku 62: 649
- James GT (1978) Inactivation of the protease inhibitor phenylmethylsulfonyl fluoride in buffers. Analyt Biochem 86: 574–579
- Jouan L, Taveau J-C, Marco S, Lallier FH, Lamy JN (2001) Occurrence of two architectural types of hexagonal bilayer hemoglobin in annelids: comparison of 3D reconstruction volumes of *Arenicola marina* and *Lumbricus terrestris* hemoglobins. J Mol Biol 305: 757–771
- Kamemoto FI, Spalding AE, Keister SM (1962) Ionic balance in blood and coelomic fluid of earthworms. Biol Bull 122: 228–231
- Krebs A, Zipper P, Vinogradov SN (1996) Lack of size and shape alteration of oxygenated and deoxygenated *Lumbricus terrestris* hemoglobin? Biochim Biophys Acta 1297: 115–118
- Krogh-Rasmussen K, Weber RE (1979) Respiratory properties of erythrocruorin (extracellular hemoglobin) in the blood of the annelid *Arenicola marina* with special reference to the influences of salinity and temperature. Ophelia 18: 151–170
- Kuchumov AR, Loo JA, Vinogradov SN (2000) Subunit distribution of calcium-binding sites in *Lumbricus terrestris* hemoglobin. J Prot Chem 19 : 139–149
- Lamy JN, Green BN, Toulmond A, Wall JS, Weber RE, Vinogradov SN (1996) Giant hexagonal bilayer hemoglobins. Chem Rev 96: 3113–3124
- Marques MB, Meirelles NC (1995) Erythrocruorin of *Glossoscolex paulistus* (Righi) (Olgochaeta, Glossoscolecidae): effects of divalent ions, acid-alkaline transition and alkali and urea denaturation. Comp Biochem Physiol 111B : 311–318
- Martin PD, Kuchumov AR, Green BN, Oliver RWA, Braswell EH, Wall JS, Vinogradov SN (1996) Mass spectrometric composition and molecular mass of *Lumbricus terrestris* hemoglobin: a refined model of its quaternary structure. J Mol Biol 255:154– 169
- Ochiai T (1983) Dissociation and oxygen equilibrium properties of earthworm (*Pheretima hilgendorfi*) hemoglobin. Arch Biochem Biophys 226: 111–117
- Ochiai T (1984) Dissociation and oxygen equilibrium properties of the extracellular hemoglobin of *Eisenia foetida*. Arch Biochem Biophys 231: 136–143
- Ochiai T, Tanaka H, Usuki I (1991) Interactions of cations to earthworm giant hemoglobin. Zool Sci (Tokyo) 8: 1134
- Ochiai T, Hoshina S, Usuki I (1993) Zinc as modulator of oxygenation function and stabilizer of quaternary structure in earthworm hemoglobin. Biochim Biophys Acta 1203: 310–314

- Oglesby LC (1978) Salt and water balance. In "Physiology of Annelids" Ed by PJ Mill, Academic Press, London, pp 555–658
- Prosser CL (1973) Comparative Animal Physiology. 3rd Ed, Saunders, Philadelphia
- Sick H, Gersonde K (1969) Method for continuous registration of O_2 -binding curves of hemoproteins by means of a diffusion chamber. Analyt Biochem 32: 362–376
- Sims RW, Gerard BM (1985) Earthworms. Synopses of the British Fauna (New Series). No.31, The Linnean Society of London and The Estuarine and Brackish-Water Sciences Association, Leiden
- Standley PR, Mainwaring MG, Gotoh T, Vinogradov SN (1988) The calcium, copper and zinc content of some annelid extracellular haemoglobins. Biochem J 249: 915–916
- Taveau J-C, Boisset N, Vinogradov SN, Lamy JN (1999) Threedimensional reconstruction of *Lumbricus terrestris* hemoglobin at 22 Å resolution: intramolecular localization of the globin and linker chains. J Mol Biol 289: 1343–1359
- Tsuneshige A, Imai K, Hori H, Tyuma I, Gotoh T (1989) Spectrophotometric, electron paramagnetic resonance and oxygen binding studies on the hemoglobin from the marine polychaete *Perinereis aibuhitensis* (Grübe): comparative physiology of hemoglobin. J Biochem 106: 406–417
- Vidugiris GJA, Harrington JP, Friedman JM, Hirsch RE (1993) Absence of ligand binding-induced tertiary changes in the multimeric earthworm *Lumbricus terrestris* hemoglobin. J Biol Chem 268: 26190–26192
- Vinogradov SN, Lugo SD, Mainwaring MG, Kapp OH, Crewe AV (1986) Bracelet protein: a quaternary structure proposed for the giant extracellular hemoglobin of *Lumbricus terrestris*. Proc Natl Acad Sci USA 83: 8034–8038
- Vinogradov SN, Sharma PK, Qabar AN, Wall JS, Westrick JA, Simmons JH, Gill SJ (1991) A dodecamer of globin chain is the principal functional subunit of the extracellular haemoglobin of *Lumbricus terrestris*. J Biol Chem 266:13091–13096
- Weber RE (1981) Cationic control of O₂ affinity in lugworm erythrocruorin. Nature 292: 386–387
- Weber RE, Olsen LF (1984) Does macromolecular surface pH explain the cation dependence of erythrocruorin oxygen affinity? Mol Physiol 6: 1–8
- Weber RE, Baldwin J (1985) Blood and erythrocruorin of the giant earthworm, *Megascolides australis*: respiratory characteristics and evidence for CO₂ facilitation of O₂ binding. Mol Physiol 7: 93–105
- Weber RE, Malte H, Braswell EH, Oliver RWA, Green BN, Sharma PK, Kuchumov A, Vinogradov SN (1995) Mass spectrometric composition, molecular mass and oxygen binding of *Macrobdella decora* hemoglobin and its tetramer and monomer subunit. J Mol Biol 251: 703–720
- Weber RE, Vinogradov SN (2001) Nonvertebrate hemoglobins: functions and molecular adaptations. Physiol Rev 81: 569–628
- Zal F, Green BN, Lallier FH, Vinogradov SN, Toulmond A (1997) Quaternary structure of the extracellular haemoglobin of the lugworm *Arenicola marina*. A multi-angle-laser-light-scattering and electrospray-ionisation-mass-spectrometry analysis. Eur. J Biochem 243: 85–92

(Received April 15, 2002 / Accepted June 20, 2002)